Mechanistic insights and antifungal assessment of (3+2) cycloaddition products of 2-diazopropane and 5-hydroxy-3-methyl-1,5-dihydropyrrol-2-one derivatives

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Basins	DAP	HMHP1	HMHP2
V(N1)	1.99		
V'(N1)	2.00		
V(N1,N2)	1.77		
V'(N1,N2)	1.73		
V(N2,C3)	1.65		
V'(N2,C3)	1.65		
V(C3)	0.47		
V'(C3)	0.47		
V(C4,C5)		1.72	
V'(C4,C5)		1.73	
V(C4',C5')			1.70
V'(C4',C5')			1.75

Table. 1S. ELF valence basin populations for reactants DAP, HMHP1, and HMHP2 in electrons e, calculated at thewB97XD/6-311++G(d,p)level of theory.



Figure 1S. (Left) ELF localization domains of the reactants plotted at an isosurface value of 0.8. Protonated basins are shown in light purple, monosynaptic basins are colored orange, disynaptic basins are colored green, and core basins are colored purple. (Centre) ELF basin attractor positions and population values for the most relevant sites. (Right) Lewis-like structures of reactants and natural atomic charge on reaction sites

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	P/TS	ΔE (kcal.mol ⁻¹)	ΔG (kcal.mol ⁻¹)	ΔH (kcal.mol ⁻¹)	ΔS (cal.K ⁻¹ .mol ⁻¹)
	P1a	-39.22	-21.07	-36.29	-55.71
	P1s	-33.61	-14.15	-30.42	-59.58
	P2a	-39.57	-20.66	-36.40	-57.62
Gas phase	P2s	-37.57	-17.53	-34.14	-60.81
	TS1a	10.91	25.54	11.47	-51.52
	TS1s	14.13	29.37	14.80	-53.36
	TS2a	8.64	24.56	9.51	-55.11
	TS2s	10.34	26.92	11.42	-60.81
	P1a	-40.38	-21.51	-37.08	-57.02
	P1s	-33.63	-13.29	-30.19	-61.86
	P2a	-39.54	-19.89	-36.28	-60.02
Et ₂ O solvent	P2s	-36.86	-16.56	-33.49	-61.97
	TS1a	10.06	25.28	10.72	-53.27
	TS1s	13.43	29.21	14.20	-54.96
	TS2a	10.75	27.04	11.67	-56.25
	TS2s	12.25	28.48	13.12	-56.25

Table. 2S. Relative energies ΔE (kcal.mol⁻¹), Gibbs free energies ΔG (kcal.mol⁻¹), relative enthalpies ΔH (kcal.mol⁻¹), and entropies ΔS (cal.mol⁻¹.K⁻¹) of the TSs and products involved in the 32CA reaction of DAP with HMHP1 in gas phase and in Et₂O solvent at 273.15 K.

Table. 3S. Relative energies ΔE (kcal.mol⁻¹), Gibbs free energies ΔG (kcal.mol⁻¹), relative enthalpies ΔH (kcal.mol⁻¹), and entropies ΔS (cal.K⁻¹.mol⁻¹) of the TSs and products involved in the 32CA reaction of DAP with HMHP2 in gas phase and in Et₂O solvent at 273.15 K.

-	P/TS	ΔE (kcal.mol ⁻¹)	ΔG (kcal.mol ⁻¹)	ΔH (kcal.mol ⁻¹)	ΔS (cal.K ⁻¹ .mol ⁻¹)
	D2a	40.10	22.25	27.54	
	r Ja	-40.19	-22.35	-37.54	-55.61
	P3s	-34.07	-14.70	-30.99	-59.65
	P4a	-40.39	-21.99	-37.50	-56.81
Gas phase	P4s	-38.14	-19.35	-35.18	-57.95
	TS3a	10.92	24.77	11.17	-49.82
	TS3s	14.08	28.18	14.34	-50.65
	TS4a	9.81	24.61	10.19	-52.81
	TS4s	10.40	26.09	10.99	-55.30
	P3a	-41.15	-22.05	-37.84	-57.80
	P3s	-33.82	-13.79	-30.37	-60.69
	P4a	-40.12	-20.80	-36.78	-58.51
Et ₂ O solvent	P4s	-37.53	-18.36	-34.26	-58.24
	TS3a	9.91	23.81	10.42	-49.05
	TS3s	13.68	28.79	14.25	-53.22
	TS4a	10.63	26.35	11.47	-54.49
	TS4s	11.38	27.64	12.35	-55.98

Structure	r (Å)	l	r (Å)	l	Structure	r (Å)	l	r (Å)	l
	N1-C4		C3-C5			N1-C4'		C3-C5'	
P1a	1.4926	0 4202	1.5499	0 5768	P3a	1.4926	0 4220	1.5498	0 5725
TS1a	2.3580	0.4202	2.2058	0.2700	TS3a	2.3552	0.4220	2.2123	0.3723
P1s	1.5003	0 4710	1.5406	0 5625	P3s	1.4999	0 4757	1.5407	0 5595
TS1s	2.2926	04/19	2.2145	0.3025	TS3s	2.2863	0.4737	2.2209	0.3383
			C2 C1						
	NI-C5		C3-C4			NI-C5 ⁷		C3-C4 ′	
P2a	1.4820	0 5769	1.5620	0 4847	P4a	1.4820	0 5746	1.5608	0 4846
TS2a	2.1091	0.5705	2.3669	0.4047	TS4a	2.1124	0.5740	2.3652	0.1010
P2s	1.4741	0.5200	1.5670	0 5194	P4s	1.4743	0 5 4 5 1	1.5669	0 5154
TS2s	2.1537	0.3390	2.3216	0.3184	TS4s	2.1450	0.3431	2.3262	0.3134

Table. 4S. Calculated bond lengths (Å) and *l* index of bond formation for the reactions of DAP with HMHP1, and HMHP2.

Table. 5S. The most significant ELF valence basin populations at the TSs have been obtained at the wB97XD/6-311++G(d,p) level of theory.

Basins	TS1a	TS1s	TS2a	TS2s
V(N1)	3.69	3.67	3.74	3.76
V(N1,N2)	1.67	1.67	1.47	1.48
V'(N1,N2)	1.43	1.40	1.51	1.48
V(N2)	1.84	1.92	1.99	2.00
V(N2,C3)	1.96	1.92	1.98	1.96
V(C3)	0.91	0.95	0.84	0.84
V(C4,C5)	3.34	3.04	3.35	3.06
V(C4)				0.30
V(C5)		0.31		



Figure 2S. (Left) ELF localization domains of the TSs of the reaction of DAP with HMHP1 plotted at an isosurface value of 0.8. Protonated basins are shown in light blue, monosynaptic basins in orange, disynaptic basins in green, and core basins in purple. (Centre) ELF basin attractor positions and population values for the important sites. (Right) Lewis-like structures of TSs

Table. 65.	The most	t significant	ELF valenc	e basin	populations	s at the	1Ss have	been	obtained	at the
wB97XD/	5-311++G	(d,p) level of	f theory.							

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Basins	TS3a	TS3s	TS4a	TS4s
V(N1)	3.69	3.68	3.73	3.75
V(N1,N2)	1.42	1.67	1.50	1.51
V'(N1,N2)	1.69	1.39	1.48	1.45
V(N2)	1.84	1.92	1.99	2.01
V(N2,C3)	1.96	1.92	1.97	1.96
V(C3)	0.91	0.96	0.84	0.85
V(C4',C5')	3.35	3.04	3.36	3.07
V(C4')				0.29
V(C5')		0.32		



Figure 3S. (Left) ELF localization domains of the TSs of the reaction of DAP with HMHP2plotted at an isosurface value of 0.8. Protonated basins are shown in light blue, monosynaptic basins in orange, disynaptic basins in green, and core basins in purple. (Centre) ELF basin attractor positions and population values for the important sites. (Right) Lewis-like structures of TSs

Phases	Ι	II	III	IV	V	VI	
Points	P1	P2	P3	P4	P5	P6	P7
d(N1,C4)	3.07	2.56	2.36	2.26	2.03	1.94	1.49
d(C3,C5)	3.12	2.48	2.21	2.06	1.80	1.73	1.55
GEDT	0.03	0.13	0.20	0.19	0.09	0.04	-0.10
V(N1,N2)	1.76	1.80	1.67	1.58	2.77	2.73	2.54
V'(N1,N2)	1.79	1.62	1.43	1.41			
V(N1)	1.90	3.81	3.69	3.65	3.68	2.92	2.72
V'(N1)	2.05					0.80	
V(N2,C3)	1.60	3.46	1.96	1.87	1.81	1.81	1.83
V'(N2,C3)	1.68						
V(C3)	0.53	0.76	0.91	0.97			
V(C4,C5)	1.69	1.60	3.34	2.90	2.22	2.15	1.96
V'(C4,C5)	1.73	1.76					
V(C4)					0.46	0.49	
V(C5)				0.46			
V(N2)			1.84	2.10	2.46	2.52	2.68
V(N1,C4)							1.80
V(C3,C5)					1.73	1.80	1.96

Table. 7S. wB97XD/6-311++G(d,p) calculated most significant electronic localization function valence basin populations, global electron density transfer, forming bond distances and of the relevant IRC points along the 32CA reaction between DAP and the HMHP1.



Figure 4S. ELF basin attractor positions of the IRC points P1-P7 of the 32CA reaction of DAP with HMHP1

Table. 8S. QTAIM parameters (in a.u.) of the	(3,-1) BCPs at the T	Ss corresponding to	the 32CA
reactions of DAP with HMHP1 and HMHP2.			

CP1		C	P2
ρ	∇2 ρ(r)	ρ	∇2 ρ(r)
(N1-	C4)	(C3-	-C5)
0.0408	0.0800	0.0626	0.0368
0.0461	0.0841	0.06201	0.0358
(N1-	C5)	(C3-	-C4)
0.0669	0.0891	0.04786	0.0411
0.0620	0.0867	0.05162	0.03897
(N1-0	C4')	(C3-	C5')
0.0410	0.0800	0.0618	0.0375
0.0467	0.0844	0.0613	0.0361
(N1-0	C 5')	(C3-	C4')
0.06642	0.0891	0.0480	0.0408
0.06282	0.0876	0.0513	0.0391
	ρ (N1- 0.0408 0.0461 (N1- 0.0669 0.0620 (N1- 0.0410 0.0467 (N1- 0.06642 0.06282	$\begin{tabular}{ c c c } \hline CP1 \\ \hline \rho & \nabla 2\rho(r) \\ \hline (N1-C4) \\ \hline 0.0408 & 0.0800 \\ 0.0461 & 0.0841 \\ \hline (N1-C5) \\ \hline 0.0669 & 0.0891 \\ 0.0669 & 0.0867 \\ \hline (N1-C4') \\ \hline 0.0410 & 0.0800 \\ 0.0467 & 0.0844 \\ \hline (N1-C5') \\ \hline 0.06642 & 0.0891 \\ 0.06282 & 0.0876 \\ \hline \end{tabular}$	$\begin{tabular}{ c c c c c } \hline C & \hline C & \hline \hline \rho & \hline \nabla 2\rho(r) & \hline \rho & \hline \hline & \hline$